ICP

GU80-1 Gray Base Coat

ICP Construction Inc.

Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Issue Date: 02/14/2023 Print Date: 02/14/2023 S.GHS.USA.EN

SECTION 1 Identification

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Product	Identifier	

Product identifier	
Product name	GU80-1 Gray Base Coat
Synonyms	Not Available
Other means of identification	Not Available

Recommended use of the chemical and restrictions on use

Relevant identified uses Waterborne Coatings

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	ICP Construction Inc.
Address	150 Dascomb Road Andover, MA 01810 United States
Telephone	1-866-667-5119 1-978-623-9987
Fax	Not Available
Website	www.icpgroup.com
Email	sds@icpgroup.com

Emergency phone number

Association / Organisation	ChemTel
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	1-813-248-0585

SECTION 2 Hazard(s) identification

Classification of the substance or mixture

NFPA 704 diamond



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification	Carcinogenicity Category 1B, Specific Target Organ Toxicity - Single Exposure Category 2, Serious Eye Damage/Eye Irritation Category 1, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Skin Corrosion/Irritation Category 2, Sensitisation (Skin) Category 1, Germ Cell Mutagenicity Category 2, Specific Target Organ Toxicity - Repeated Exposure Category 1
Label elements	
Hazard pictogram(s)	
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Signal word	Danger
Hazard statement(s)	
H350	May cause cancer.

H371	May cause damage to organs.
H318	Causes serious eye damage.
H335	May cause respiratory irritation.
H315	Causes skin irritation.
H317	May cause an allergic skin reaction.
H341	Suspected of causing genetic defects.
H372	Causes damage to organs through prolonged or repeated exposure.

Hazard(s) not otherwise classified

Not Applicable

Precautionary statement(s) General

P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.
P103	Read label before use.

Precautionary statement(s) Prevention

P201	Obtain special instructions before use.
P260	Do not breathe mist/vapours/spray.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P270	Do not eat, drink or smoke when using this product.
P202	Do not handle until all safety precautions have been read and understood.
P264	Wash all exposed external body areas thoroughly after handling.
P272	Contaminated work clothing must not be allowed out of the workplace.

Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P308+P313	IF exposed or concerned: Get medical advice/ attention.
P310	Immediately call a POISON CENTER/doctor/physician/first aider.
P308+P311	IF exposed: Call a POISON CENTER or doctor/physician.
P312	Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.
P314	Get medical advice/attention if you feel unwell.
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.
P302+P352	IF ON SKIN: Wash with plenty of water and soap.
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P332+P313	If skin irritation occurs: Get medical advice/attention.
P362+P364	Take off contaminated clothing and wash it before reuse.

Precautionary statement(s) Storage

P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

P501

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
14808-60-7	1-5	silica crystalline - quartz
65997-15-1	15-40	portland cement
13397-24-5	1-5	gypsum
1317-65-3	1-5	limestone
65996-69-2	1-5	blast furnace slag

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

SECTION 4 First-aid measures

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay.
Ingestion	 Immediately give a glass of water. First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

Treat symptomatically

For acute or short term repeated exposures to iron and its derivatives:

- Always treat symptoms rather than history.
- In general, however, toxic doses exceed 20 mg/kg of ingested material (as elemental iron) with lethal doses exceeding 180 mg/kg.
- Control of iron stores depend on variation in absorption rather than excretion. Absorption occurs through aspiration, ingestion and burned skin.
- + Hepatic damage may progress to failure with hypoprothrombinaemia and hypoglycaemia. Hepatorenal syndrome may occur.
- Iron intoxication may also result in decreased cardiac output and increased cardiac pooling which subsequently produces hypotension.
- Serum iron should be analysed in symptomatic patients. Serum iron levels (2-4 hrs post-ingestion) greater that 100 ug/dL indicate poisoning with levels, in excess of 350 ug/dL, being potentially serious. Emesis or lavage (for obtunded patients with no gag reflex)are the usual means of decontamination.
- Activated charcoal does not effectively bind iron.
- Catharsis (using sodium sulfate or magnesium sulfate) may only be used if the patient already has diarrhoea.
- Deferoxamine is a specific chelator of ferric (3+) iron and is currently the antidote of choice. It should be administered parenterally. [Ellenhorn and Barceloux: Medical Toxicology]

For acute or short term repeated exposures to dichromates and chromates:

- Absorption occurs from the alimentary tract and lungs.
- The kidney excretes about 60% of absorbed chromate within 8 hours of ingestion. Urinary excretion may take up to 14 days.
- Establish airway, breathing and circulation. Assist ventilation.
- Induce emesis with Ipecac Syrup if patient is not convulsing, in coma or obtunded and if the gag reflex is present.
- Otherwise use gastric lavage with endotracheal intubation.
- Fluid balance is critical. Peritoneal dialysis, haemodialysis or exchange transfusion may be effective although available data is limited.
- British Anti-Lewisite, ascorbic acid, folic acid and EDTA are probably not effective.
- There are no antidotes.
- Primary irritation, including chrome ulceration, may be treated with ointments comprising calcium-sodium-EDTA. This, together with the use of frequently renewed dressings, will ensure rapid healing of any ulcer which may develop.

The mechanism of action involves the reduction of Cr (VI) to Cr(III) and subsequent chelation; the irritant effect of Cr(III)/ protein complexes is thus avoided. [ILO Encyclopedia]

[Ellenhorn and Barceloux: Medical Toxicology]

- Manifestation of aluminium toxicity include hypercalcaemia, anaemia, Vitamin D refractory osteodystrophy and a progressive encephalopathy (mixed dysarthria-apraxia of speech, asterixis, tremulousness, myoclonus, dementia, focal seizures). Bone pain, pathological fractures and proximal myopathy can occur.
- Symptoms usually develop insidiously over months to years (in chronic renal failure patients) unless dietary aluminium loads are excessive.
- Serum aluminium levels above 60 ug/ml indicate increased absorption. Potential toxicity occurs above 100 ug/ml and clinical symptoms are present when levels exceed 200 ug/ml.
- Deferoxamine has been used to treat dialysis encephalopathy and osteomalacia. CaNa2EDTA is less effective in chelating aluminium.

[Ellenhorn and Barceloux: Medical Toxicology]

- For acute or short-term repeated exposures to highly alkaline materials:
 - Respiratory stress is uncommon but present occasionally because of soft tissue edema.
 - Unless endotracheal intubation can be accomplished under direct vision, cricothyroidotomy or tracheotomy may be necessary.
 - Oxygen is given as indicated.
 - The presence of shock suggests perforation and mandates an intravenous line and fluid administration.

Damage due to alkaline corrosives occurs by liquefaction necrosis whereby the saponification of fats and solubilisation of proteins allow deep penetration into the tissue.

Alkalis continue to cause damage after exposure

INGESTION:

Milk and water are the preferred diluents

- No more than 2 glasses of water should be given to an adult.
- Neutralising agents should never be given since exothermic heat reaction may compound injury.
- * Catharsis and emesis are absolutely contra-indicated.
- * Activated charcoal does not absorb alkali.
- * Gastric lavage should not be used.

Supportive care involves the following:

- Withhold oral feedings initially.
- If endoscopy confirms transmucosal injury start steroids only within the first 48 hours.
- Carefully evaluate the amount of tissue necrosis before assessing the need for surgical intervention
- Patients should be instructed to seek medical attention whenever they develop difficulty in swallowing (dysphagia).
- SKIN AND EYE:

Injury should be irrigated for 20-30 minutes.

Eye injuries require saline. [Ellenhorn & Barceloux: Medical Toxicology]

SECTION 5 Fire-fighting measures

Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
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Special protective equipment and precautions for fire-fighters

Fire Fighting	 When silica dust is dispersed in air, firefighters should wear inhalation protection as hazardous substances from the fire may be adsorbed on the silica particles. When heated to extreme temperatures, (>1700 deg.C) amorphous silica can fuse. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire.
Fire/Explosion Hazard	 Under certain conditions the material may become combustible because of the ease of ignition which occurs after the material reaches a high specific area ratio (thin sections, fine particles, or molten states). However, the same material in massive solid form is comparatively difficult to ignite. When silica dust is dispersed in air, firefighters should wear inhalation protection as hazardous substances from the fire may be adsorbed on the silica particles. When heated to extreme temperatures, (>1700 deg.C) amorphous silica can fuse. Decomposition may produce toxic fumes of: silicon dioxide (SiO2) metal oxides When aluminium oxide dust is dispersed in air, firefighters should wear protection against inhalation of dust particles, which can also contain hazardous substances from the fire absorbed on the alumina particles. May emit poisonous fumes.

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. DO NOT allow clothing wet with material to stay in contact with skin

Conditions for safe storage, including any incompatibilities

Suitable container	 Polyethylene or polypropylene container. Packing as recommended by manufacturer.
Storage incompatibility	 Derivative of electropositive metal. For aluminas (aluminium oxide): Incompatible with hot chlorinated rubber. In the presence of chlorine trifluoride may react violently and ignite. Calcium oxide: reacts violently with water, evolving high quantities of heat reacts violently, with possible ignition or explosion, with acids, anilinium perchlorate, bromine pentafluoride, chlorine trifluoride, fluorine, hydrogen fluoride, hydrogen sulfide, hydrogen trisulfide, isopropyl isocyanide dichloride, light metals, lithium, magnesium, powdered aluminium, phosphorus, potassium, sulfur trioxide increase the explosive sensitivity of azides, nitroalkanes (e.g. nitroethane, nitromethane, 1-nitropropane etc.) is incompatible with boric acid, boron trifluoride, carbon dioxide, ethanol, halogens (such as fluorine), metal halides, phosphorus pentoxide, selenium oxychloride, sulfur dioxide and many organic materials

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	Calcium sulfate:
	 reacts violently with reducing agents, acrolein, alcohols, chlorine trifluoride, diazomethane, ethers, fluorine, hydrazine, hydrazinium perchlorate, hydrogen peroxide, finely divided aluminium or magnesium, peroxyfuroic acid, red phosphorus, sodium acetylide sensitises most organic azides which are unstable shock- and heat- sensitive explosives
	may form explosive materials with 1.3-di(5-tetrazolv))triazene
	is incompatible with glycidol, isopropyl chlorocarbonate, nitrosyl perchlorate, sodium borohydride
	is hydroscopic: reacts with water to form gypsum and Plaster of Paris
	For iron oxide (ferric oxide):
	Avoid storage with aluminium, calcium hypochlorite and ethylene oxide.
	 Risk of explosion occurs following reaction with powdered aluminium, calcium silicide, ethylene oxide (polymerises), carbon monoxide, magnesium and perchlorates.
	The substance may be or contains a "metalloid"
	The following elements are considered to be metalloids; boron, silicon, germanium, arsenic, antimony, tellurium and (possibly) polonium
	The electronegativities and ionisation energies of the metalloids are between those of the metals and nonmetals, so the metalloids exhibit
	characteristics of both classes. The reactivity of the metalloids depends on the element with which they are reacting.
	WARNING: Avoid or control reaction with peroxides. All transition metal peroxides should be considered as potentially explosive.
	Silicas:
	react with hydrofluoric acid to produce silicon tetrafluoride gas
	react with xenon hexafluoride to produce explosive xenon trioxide
	reacts exothermically with oxygen difluoride, and explosively with chlorine trifluoride (these halogenated materials are not commonplace industrial materials) and other fluorine-containing compounds
	may react with fluorine, chlorates
	are incompatible with strong oxidisers, manganese trioxide, chlorine trioxide, strong alkalis, metal oxides, concentrated orthophosphoric acid, vinyl acetate
	may react vigorously when heated with alkali carbonates.
	Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.
	Avoid contact with copper, aluminium and their alloys.
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SECTION 6 Exposure contr	ois / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

INGREDIENT DATA						
Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Limits (PELs) Table Z-1	silica crystalline - quartz	Quartz - respirable	0.05 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-3	silica crystalline - quartz	Silica: Crystalline: Quartz (Respirable)	10 (%SiO2+2) mg/m3 / 250 (%SiO2+5) mppcf	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	silica crystalline - quartz	Silica, crystalline (as respirable dust)	0.05 mg/m3	Not Available	Not Available	Ca; See Appendix A
US OSHA Permissible Exposure Limits (PELs) Table Z-1	portland cement	Portland cement- Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	portland cement	Portland cement- Total dust	15 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-3	portland cement	Silicates (less than 1% crystalline silica): Portland cement	50 mppcf	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	portland cement	Portland cement - respirable	5 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	portland cement	Portland cement - total	10 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	gypsum	Gypsum- Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	gypsum	Calcium sulfate- Total dust	15 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	gypsum	Calcium sulfate- Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	gypsum	Gypsum- Total dust	15 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	gypsum	Gypsum - total	10 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	gypsum	Gypsum - respirable	5 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	gypsum	Calcium sulfate - total	10 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	gypsum	Calcium sulfate - respirable	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Calcium Carbonate- Total dust	15 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Calcium Carbonate- Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Limestone- Total dust	15 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Limestone- Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available

Source	Ingredient	Material name		TWA		STEL	Peak	Notes
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Marble- Total dust		15 mg/m3		Not Available	Not Available	Not Available
US OSHA Permissible Exposure Limits (PELs) Table Z-1	limestone	Marble- Respirable fraction	I	5 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Calcium carbonate - total		10 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Calcium carbonate - total		10 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Limestone - respirable		5 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Marble - total		10 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Marble - respirable		5 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Limestone - total		10 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Calcium carbonate - respira	able	5 mg/m3		Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	limestone	Calcium carbonate - respira	able	5 mg/m3		Not Available	Not Available	Not Available
Emergency Limits								
Ingredient	TEEL-1		TEEL-2			TEEL-3		
silica crystalline - quartz	0.075 mg/m3		33 mg/m3			200 mg/m3		
limestone	45 mg/m3		210 mg/m3			1,300 mg/m3		
Ingredient	Original IDLH	Original IDLH			Revised	IDLH		
silica crystalline - quartz	25 mg/m3 / 50 mg/n	25 mg/m3 / 50 mg/m3			Not Available			
portland cement	5,000 mg/m3				Not Available			

Not Available

Not Available

Not Available

Exposure controls

blast furnace slag

gypsum limestone

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.
Personal protection	
Eye and face protection	 Safety glasses with side shields. Chemical goggles.
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. Neoprene rubber gloves
Body protection	See Other protection below
Other protection	 Employees working with confirmed human carcinogens should be provided with, and be required to wear, clean, full body protective clothing (smocks, coveralls, or long-sleeved shirt and pants), shoe covers and gloves prior to entering the regulated area. [AS/NZS ISO 6529:2006 or national equivalent] Employees engaged in handling operations involving carcinogens should be provided with, and required to wear and use half-face filter-type respirators with filters for dusts, mists and fumes, or air purifying carcinogens, employees should be required to remove and leave protective clothing and equipment at the point of exit and at the last exit of the day, to place used clothing and equipment in impervious containers at the point of exit for purposes of decontamination or disposal. The contents of such impervious containers must be identified with suitable labels. Overalls. P.V.C apron.

Respiratory protection

If inhalation risk above the TLV exists, wear approved dust respirator.

Use respirators with protection factors appropriate for the exposure level.

• Up to 5 X TLV, use valveless mask type; up to 10 X TLV, use 1/2 mask dust respirator

Not Available

Not Available

Not Available

▶ Up to 50 X TLV, use full face dust respirator or demand type C air supplied respirator

- ▶ Up to 500 X TLV, use powered air-purifying dust respirator or a Type C pressure demand supplied-air respirator
- Over 500 X TLV wear full-face self-contained breathing apparatus with positive pressure mode or a combination respirator with a Type C positive pressure supplied-air full-face respirator and an auxiliary self-contained breathing apparatus operated in pressure demand or other positive pressure mode

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Not Available		
	1		1
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation may result in ulcers or sores of the lining of the nose (nasal mucosa), and lung damage. The material has NOT been classified by EC Directives or other classification systems as "harmful by inhalation". This is because of the lack of corroborating animal or human evidence. Inhalation of dusts, generated by the material during the course of normal handling, may be damaging to the health of the individual. Effects on lungs are significantly enhanced in the presence of respirable particles.
Ingestion	Chromate salts are corrosive and produce cellular damage to tissue. Ingestion may produce inflammation of the digestive tract, nausea, vomiting and abdominal pain. The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence. Accidental ingestion of the material may be damaging to the health of the individual.
Skin Contact	This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. Though considered non-harmful, slight irritation may result from contact because of the abrasive nature of the aluminium oxide particles. Thus it may cause itching and skin reaction and inflammation.

	Four students received severe hand burns whilst making moulds of the plaster known as "Stone" was a special form of calcium sulfate hemith strength to the moulds. Handling wet cement can cause dermatitis. Cement when wet is quite contact dermatitis since it may cause drying and defatting of the skin infections of lesions and penetration by soluble salts. Skin contact may result in severe irritation particularly to broken skin. Open cuts, abraded or irritated skin should not be exposed to this ma Entry into the blood-stream, through, for example, cuts, abrasions or prior to the use of the material and ensure that any external damage is	eir hands with dental plaster substituted for Plaster of Paris. The dental ydrate containing alpha-hemihydrate crystals that provide high compression e alkaline and this alkali action on the skin contributes strongly to cement which is followed by hardening, cracking, lesions developing, possible Ulceration known as "chrome ulcers" may develop. terial lesions, may produce systemic injury with harmful effects. Examine the skin s suitably protected.
Eye	If applied to the eyes, this material causes severe eye damage.	
Chronic	Long-term exposure to respiratory irritants may result in airways disea Strong evidence exists that this substance may cause irreversible mu Skin contact with the material is more likely to cause a sensitisation re There is sufficient evidence to suggest that this material directly cause Toxic: danger of serious damage to health by prolonged exposure the This material can cause serious damage if one is exposed to it for lon produce severe defects. Substance accumulation, in the human body, may occur and may cau. Animal testing shows long term exposure to aluminium oxides may ca smaller the size, the greater the tendencies of causing harm. Red blood cells and rabbit alveolar macrophages exposed to calcium in another. Both studies showed the substance to be more cytotoxic th Cement contact dermatitis (CCD) may occur when contact shows an to soluble chromates (chromate compounds) present in trace amount Crystalline silicas activate the inflammatory response of white blood c silicas reduces lung capacity and predisposes to chest infections. Amorphous silicas generally are less hazardous than crystalline silicas cooling. Inhalation of dusts containing crystalline silicas may lead to s Soluble silicates do not exhibit sensitizing potential. Testing in bacteria mutations or birth defects. Chromium (III) is an essential trace mineral. Chronic exposure to chroc fluid in the lungs, and adverse effects on white blood cells, and also ir Chronic excessive intake of iron have been associated with damage t over iron are at an increased risk. Overexposure to the breathable dust may cause coughing, wheezing, include decreased vital lung capacity and chest infections.	ase, involving difficulty breathing and related whole-body problems. tations (though not lethal) even following a single exposure. saction in some persons compared to the general population. es cancer in humans. ough inhalation, in contact with skin and if swallowed. g periods. It can be assumed that it contains a substance which can use some concern following repeated or long-term occupational exposure. ause lung disease and cancer, depending on the size of the particle. The silicate insulation materials in vitro showed haemolysis in one study but not han titanium dioxide but less toxic than asbestos. allergic response, which may progress to sensitisation. Sensitisation is due s is nome cements and cement products. tells after they injure the lung epithelium. Chronic exposure to crystalline s, but the former can be converted to the latter on heating and subsequent ilicosis, a disabling lung disease that may take years to develop. al and animal experiments have not shown any evidence of them causing omium (III) irritates the airways, malnourishes the liver and kidneys, causes ncreases the risk of developing lung cancer. o the liver and pancreas. People with a genetic disposition to poor control n difficulty in breathing and impaired lung function. Chronic symptoms may
	ΤΟΧΙΟΙΤΥ	IRRITATION
GU80-1 Gray Base Coat	Not Available	Not Available
	тохісіту	IRRITATION
silica crystalline - quartz	Oral (Rat) LD50: 500 mg/kg ^[2]	Not Available

	Oral (Rat) LD50: 500 mg/kg ^[2]	Not Available
mentioned assessed	ΤΟΧΙΟΙΤΥ	IRRITATION
portiand cement	Not Available	Not Available
	ΤΟΧΙΟΙΤΥ	IRRITATION
gypsum	Inhalation(Rat) LC50: >3.26 mg/l4h ^[1]	Not Available
	Oral (Rat) LD50: >1581 mg/kg ^[1]	
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Oral (Rat) LD50: 6450 mg/kg ^[2]	Eye: no adverse effect observed (not irritating) ^[1]
limestone		Skin (rabbit): 500 mg/24h-moderate
		Skin: no adverse effect observed (not irritating) ^[1]
	тохісіту	IRRITATION
blast furnace slag	dermal (rat) LD50: >4000 mg/kg ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
	Inhalation(Rat) LC50: >5.235 mg/L4h ^[1]	Skin: no adverse effect observed (not irritating) ^[1]
	Oral (Rat) LD50: >2000 mg/kg ^[1]	
Legend:	 Value obtained from Europe ECHA Registered Substar specified data extracted from RTECS - Register of Toxic E 	nces - Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise Effect of chemical Substances

GU80-1 Gray Base Coat	Laboratory (in vitro) and animal studies show, exposure to the material may result in a possible risk of irreversible effects, with the possibility of producing mutation.
SILICA CRYSTALLINE - QUARTZ	WARNING: For inhalation exposure <u>ONLY</u> : This substance has been classified by the IARC as Group 1: CARCINOGENIC TO HUMANS The International Agency for Research on Cancer (IARC) has classified occupational exposures to respirable (<5 um) crystalline silica as being carcinogenic to humans . This classification is based on what IARC considered sufficient evidence from epidemiological studies of humans for the carcinogenicity of inhaled silica in the forms of quartz and cristobalite.
GYPSUM	Gypsum (calcium sulfate dehydrate) irritates the skin, eye, mucous membranes, and airways. A series of studies involving Gypsum industry workers in Poland reported chronic, non-specific airways diseases.

LIMESTONE	Eye (rabbit) 0.75: mg/24h - No evidence of carcinogenic properties. No evidence of mutagenic or teratogenic effects. The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis. The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.		
GU80-1 Gray Base Coat & PORTLAND CEMENT & GYPSUM & BLAST FURNACE SLAG	Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound.		
GU80-1 Gray Base Coat & PORTLAND CEMENT	The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema.		
GU80-1 Gray Base Coat & BLAST FURNACE SLAG	For silica amorphous: Derived No Adverse Effects Level (NOAEL) in the range of 1000 mg/kg/d. In humans, synthetic amorphous silica (SAS) is essentially non-toxic by mouth, skin or eyes, and by inhalation. Epidemiology studies show little evidence of adverse health effects due to SAS.		
PORTLAND CEMENT & GYPSUM & BLAST FURNACE SLAG	No significant acute toxicological data identified in literature search.		
Acute Toxicity	×	Carcinogenicity	¥
Skin Irritation/Corrosion	×	Reproductivity	×
Serious Eye Damage/Irritation	×	STOT - Single Exposure	×
Respiratory or Skin sensitisation	✓	STOT - Repeated Exposure	*
Mutagenicity	◆	Aspiration Hazard	×
		Legend: 🗙 – Data either r 🗸 – Data availab	ot available or does not fill the criteria for classification le to make classification

SECTION 12 Ecological information

	Endpoint	Test Duration (hr)	Species	Value	Source
GU80-1 Gray Base Coat	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
silica crystalline - quartz	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
portland cement	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	72h	Algae or other aquatic plants	>79mg/l	2
gypsum	LC50	96h	Fish	>79mg/l	2
	NOEC(ECx)	0.25h	Fish	75mg/l	4
	EC50	96h	Algae or other aquatic plants	3200mg/l	4
	Endpoint	Test Duration (hr)	Species	Value	Source
	NOEC(ECx)	1h	Fish	4-320mg/l	4
limestone	LC50	96h	Fish	>165200mg/L	4
	EC50	72h	Algae or other aquatic plants	>14mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
blast furnace slag	NOEC(ECx)	72h	Algae or other aquatic plants	>=100mg/l	2
	EC50	72h	Algae or other aquatic plants	>100mg/l	2
	LC50	96h	Fish	>100000mg/L	2
	EC50	48h	Crustacea	>100mg/l	2

For Metal:

Atmospheric Fate - Metal-containing inorganic substances generally have negligible vapour pressure and are not expected to partition to air.

Environmental Fate: Environmental processes, such as oxidation, the presence of acids or bases and microbiological processes, may transform insoluble metals to more soluble ionic forms.

For Chromium: Chromium is poorly absorbed by cells found in microorganisms, plants and animals. Hexavalent chromate anions are readily transported into cells and toxicity is closely linked to the higher oxidation state.

For chromium:

Aquatic Fate - Most chromium released into water will be deposited in the sediment. A small percentage of chromium can be found in soluble and insoluble forms with soluble chromium making up a very small percentage of the total chromium.

For Amorphous Silica: Amorphous silica is chemically and biologically inert. It is not biodegradable.

LOW (KOC = 6.124)

For Silica:

gypsum

Environmental Fate: Most documentation on the fate of silica in the environment concerns dissolved silica, in the aquatic environment, regardless of origin, (man-made or natural), or structure, (crystalline or amorphous).

Terrestrial Fate: Silicon makes up 25.7% of the Earth's crust, by weight, and is the second most abundant element, being exceeded only by oxygen.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
gypsum	HIGH	HIGH

Bioaccumulative potential	
Ingredient	Bioaccumulation
gypsum	LOW (LogKOW = -2.2002)
Mobility in soil	
Ingredient	Mobility

SECTION 13 Disposal considerations

Waste treatment methods		
Product / Packaging disposal	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible. Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. Recycle wherever possible. Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified. 	

SECTION 14 Transport information

Labels Required	
Marine Pollutant	NO

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
silica crystalline - quartz	Not Available
portland cement	Not Available
gypsum	Not Available
limestone	Not Available
blast furnace slag	Not Available

Transport in bulk in accordance with the IGC Code

Product name	Ship Type
silica crystalline - quartz	Not Available
portland cement	Not Available
gypsum	Not Available
limestone	Not Available
blast furnace slag	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

silica crystalline - quartz is found on the following regulatory lists

Chemical Footprint Project - Chemicals of High Concern List	US NIOSH Carcinogen List
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC	US NIOSH Recommended Exposure Limits (RELs)
Monographs	US OSHA Carcinogens Listing
Monographs - Group 1: Carcinogenic to humans	US OSHA Permissible Exposure Limits (PELs) Table Z-1
US - California Proposition 65 - Carcinogens	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - California Safe Drinking Water and Toxic Enforcement Act of 1986 - Proposition 65 List	US TSCA Chemical Substance Inventory - Interim List of Active Substances
US - Massachusetts - Right To Know Listed Chemicals US DOE Temporary Emergency Exposure Limits (TEELs)	
US National Toxicology Program (NTP) 15th Report Part A Known to be Human	
Carcinogens	
portland cement is found on the following regulatory lists	
US - Massachusetts - Right To Know Listed Chemicals	US OSHA Permissible Exposure Limits (PELs) Table Z-3
US NIOSH Recommended Exposure Limits (RELs)	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US OSHA Permissible Exposure Limits (PELs) Table Z-1	US TSCA Chemical Substance Inventory - Interim List of Active Substances
gypsum is found on the following regulatory lists	
US - Massachusetts - Right To Know Listed Chemicals	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US NIOSH Recommended Exposure Limits (RELs)	US TSCA Chemical Substance Inventory - Interim List of Active Substances
US OSHA Permissible Exposure Limits (PELs) Table Z-1	
limestone is found on the following regulatory lists	
US - Massachusetts - Right To Know Listed Chemicals	US OSHA Permissible Exposure Limits (PELs) Table Z-1
US DOE Temporary Emergency Exposure Limits (TEELs)	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US NIOSH Recommended Exposure Limits (RELs)	US TSCA Chemical Substance Inventory - Interim List of Active Substances
blast furnace slag is found on the following regulatory lists	
	UC TOOA Observiced Substance Javanteen, Jatania List of Asther Outstances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	US ISCA Chemical Substance Inventory - Interim List of Active Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	US ISCA Chemical Substance Inventory - Interim List of Active Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations	US ISCA Chemical Substance Inventory - Interim List of Active Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA)	US ISCA Chemical Substance Inventory - Interim List of Active Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories	US ISCA Chemical Substance Inventory - Interim List of Active Substances
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids)	No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure	No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive	No No No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating	No No No No No No No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid)	No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas	US ISCA Chemical Substance Inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal	US ISCA Chemical Substance Inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas)	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive	US ISCA Chemical Substance Inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure)	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity	US I SCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization	US ISCA Chemical Substance inventory - Interim List of Active Substances No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization Specific target organ toxicity (single or repeated exposure)	US ISCA Chemical Substance inventory - Interim List of Active Substances No Yes No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization Specific target organ toxicity (single or repeated exposure) Assiration Hazard	US ISCA Chemical Substance inventory - Interim List of Active Substances No Yes No
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization Specific target organ toxicity (single or repeated exposure) Aspiration Hazard Carcel mutagenicity	US ISCA Chemical Substance inventory - Interim List of Active Substances No Yes Yes Yes No Yes No Yes No No Yes No Yes No Yes No Yes
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization Serious eye damage or eye irritation Specific target organ toxicity (single or repeated exposure) Aspiration Hazard Germ cell mutagenicity	US ISCA Chemical Substance inventory - Interim List of Active Substances No Yes Yes No Yes No Yes No Yes No Yes No Yes No No </td
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory Federal Regulations Superfund Amendments and Reauthorization Act of 1986 (SARA) Section 311/312 hazard categories Flammable (Gases, Aerosols, Liquids, or Solids) Gas under pressure Explosive Self-heating Pyrophoric (Liquid or Solid) Pyrophoric Gas Corrosive to metal Oxidizer (Liquid, Solid or Gas) Organic Peroxide Self-reactive In contact with water emits flammable gas Combustible Dust Carcinogenicity Acute toxicity (any route of exposure) Reproductive toxicity Skin Corrosion or Irritation Respiratory or Skin Sensitization Serious eye damage or eye irritation Specific target organ toxicity (single or repeated exposure) Aspiration Hazard Simple Asphyxiant	US ISCA Chemical Substance inventory - Interim List of Active Substances No Yes Yes No Yes No Yes No Yes No Yes No No No No No No No No

Hazards Not Otherwise Classified

US. EPA CERCLA Hazardous Substances and Reportable Quantities (40 CFR 302.4)

None Reported

State Regulations

US. California Proposition 65

MARNING: This product can expose you to chemicals including silica crystalline - quartz, which is known to the State of California to cause cancer. For more information, go to www.P65Warnings.ca.gov.

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia	Yes	

National Inventory	Status
Non-Industrial Use	
Canada - DSL	Yes
Canada - NDSL	No (silica crystalline - quartz; portland cement; gypsum; blast furnace slag)
China - IECSC	No (blast furnace slag)
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	No (portland cement; blast furnace slag)
Korea - KECI	No (blast furnace slag)
New Zealand - NZIoC	Yes
Philippines - PICCS	No (portland cement; blast furnace slag)
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	No (blast furnace slag)
Vietnam - NCI	Yes
Russia - FBEPH	No (blast furnace slag)
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

SECTION 16 Other information

Revision Date	02/14/2023
Initial Date	02/15/2023

CONTACT POINT

PLEASE NOTE THAT TITANIUM DIOXIDE IS NOT PRESENT IN CLEAR OR NEUTRAL BASES

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit。 IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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end of SDS