

PD Stain Color Vial Brandy

ICP Building Solutions Group/Pli-Dek

Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Issue Date: **04/21/2020** Print Date: **04/21/2020** S.GHS.USA.EN

SECTION 1 IDENTIFICATION

Product Identifier

Product name	PD Stain Color Vial Brandy
Synonyms	Not Available
Other means of identification	Not Available

Recommended use of the chemical and restrictions on use

Relevant identified uses Col

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	ICP Building Solutions Group/Pli-Dek
Address	4565 W. Watkins Street Phoenix AZ Not applicable
Telephone	623-435-2277
Fax	Not Available
Website	www.ICPGROUP.com
Email	Not Available

Emergency phone number

37, F	
Association / Organisation	ChemTel
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	1-813-248-0585

SECTION 2 HAZARD(S) IDENTIFICATION

Classification of the substance or mixture

NFPA 704 diamond



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification

Skin Sensitizer Category 1, Specific target organ toxicity - repeated exposure Category 1

Label elements

Hazard pictogram(s)





SIGNAL WORD

DANGER

Hazard statement(s)

H317	May cause an allergic skin reaction.
H372	Causes damage to organs through prolonged or repeated exposure.

Hazard(s) not otherwise classified

Not Applicable

Version No: **1.1** Page **2** of **10** Issue Date: **04/21/2020**

PD Stain Color Vial Brandy

Print Date: 04/21/2020

Precautionary statement(s) General

P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.

Precautionary statement(s) Prevention

P202	Do not handle until all Safety Precautions have been read and understood.
P260	Do not breathe mist/vapours/spray.
P280	Wear protective gloves/protective clothing/eye protection/face protection.
P272	Contaminated work clothing should not be allowed out of the workplace.
P264	Wash thoroughly after handling.
P270	Do not eat, drink or smoke when using this product.

Precautionary statement(s) Response

P302+P352	IF ON SKIN: Wash with plenty of water.	
P333+P313	IF SKIN irritation or rash occurs; get medical advice/attention.	
P363	Wash contaminated clothing before reuse.	
P314	Get medical advice/attention if you feel unwell.	

Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
14807-96-6	5-20	talc
147-14-8	.5-5	C.I. Pigment Blue 15
55406-53-6	.1-5	3-iodo-2-propynyl butyl carbamate
1309-37-1	10-20	ferric oxide
1333-86-4	25-5	carbon black

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

SECTION 4 FIRST-AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with eyes: • Wash out immediately with water. • If irritation continues, seek medical attention. • Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
Ingestion	 Immediately give a glass of water. First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

for copper intoxication:

- ▶ Unless extensive vomiting has occurred empty the stomach by lavage with water, milk, sodium bicarbonate solution or a 0.1% solution of potassium ferrocyanide (the resulting copper ferrocyanide is insoluble).
- ► Administer egg white and other demulcents.
- Maintain electrolyte and fluid balances.
- ▶ Morphine or meperidine (Demerol) may be necessary for control of pain.

Version No: 1.1 Issue Date: 04/21/2020 Page 3 of 10

PD Stain Color Vial Brandy

Print Date: 04/21/2020

- If symptoms persist or intensify (especially circulatory collapse or cerebral disturbances, try BAL intramuscularly or penicillamine in accordance with the supplier's recommendations
- ▶ Treat shock vigorously with blood transfusions and perhaps vasopressor amines.
- F If intravascular haemolysis becomes evident protect the kidneys by maintaining a diuresis with mannitol and perhaps by alkalinising the urine with sodium bicarbonate.
- Fit is unlikely that methylene blue would be effective against the occassional methaemoglobinemia and it might exacerbate the subsequent haemolytic episode.
- ▶ Institute measures for impending renal and hepatic failure.

[GOSSELIN, SMITH & HODGE: Commercial Toxicology of Commercial Products]

- ▶ A role for activated charcoals for emesis is, as yet, unproven.
- In severe poisoning CaNa2EDTA has been proposed.

[ELLENHORN & BARCELOUX: Medical Toxicology]

For acute or short term repeated exposures to iron and its derivatives

- Always treat symptoms rather than history
- In general, however, toxic doses exceed 20 mg/kg of ingested material (as elemental iron) with lethal doses exceeding 180 mg/kg.
- Control of iron stores depend on variation in absorption rather than excretion. Absorption occurs through aspiration, ingestion and burned skin.
- Hepatic damage may progress to failure with hypoprothrombinaemia and hypoglycaemia. Hepatorenal syndrome may occur.
- Iron intoxication may also result in decreased cardiac output and increased cardiac pooling which subsequently produces hypotension.
- Serum iron should be analysed in symptomatic patients. Serum iron levels (2-4 hrs post-ingestion) greater that 100 ug/dL indicate poisoning with levels, in excess of 350 ug/dL, being potentially serious. Emesis or lavage (for obtunded patients with no gag reflex) are the usual means of decontamination.
- ► Activated charcoal does not effectively bind iron.
- Catharsis (using sodium sulfate or magnesium sulfate) may only be used if the patient already has diarrhoea.
- ▶ Deferoxamine is a specific chelator of ferric (3+) iron and is currently the antidote of choice. It should be administered parenterally. [Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 FIRE-FIGHTING MEASURES

Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.	
Special protective equipment and precautions for fire-fighters		
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. 	
Fire/Explosion Hazard	► Non combustible. ► Not considered a significant fire risk, however containers may burn. Decomposition may produce toxic fumes of: hydrogen iodide silicon dioxide (SiO2) metal oxides May emit corrosive fumes	

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

5. F.	
Minor Spills	 Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes.
Major Spills	Moderate hazard. ► Clear area of personnel and move upwind.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling	
Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. DO NOT allow clothing wet with material to stay in contact with skin
Other information	

Conditions for safe storage, including any incompatibilities

Suitable container Polyethylene or polypropylene container. Packing as recommended by manufacturer.
--

Version No: 1.1 Page 4 of 10 Issue Date: 04/21/2020

PD Stain Color Vial Brandy

Print Date: 04/21/2020

Storage incompatibility

Inorganic derivative of Group 11 metal. Derivative of electropositive metal. For iron oxide (ferric oxide):

- ▶ Avoid storage with aluminium, calcium hypochlorite and ethylene oxide.
- Risk of explosion occurs following reaction with powdered aluminium, calcium silicide, ethylene oxide (polymerises), carbon monoxide, magnesium and perchlorates.

 • WARNING: Avoid or control reaction with peroxides. All transition metal peroxides should be considered as potentially explosive.

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US NIOSH Recommended Exposure Limits (RELs)	talc	Massive talc, Soapstone silicate, Steatite	6 (total), 3 (resp) mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	talc	Hydrous magnesium silicate, Steatite talc	2 (resp) mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z3	talc	Silicates: Soapstone	20 mppcf	Not Available	Not Available	(Name ((less than 1% crystalline silica)))
US OSHA Permissible Exposure Levels (PELs) - Table Z3	talc	Silicates: Talc	Not Available	Not Available	Not Available	(Name ((less than 1% crystalline silica); (containing asbestos) Use asbestos limit))
US OSHA Permissible Exposure Levels (PELs) - Table Z3	talc	Silicates: Talc	20 mppcf	Not Available	Not Available	(Name ((less than 1% crystalline silica); (not containing asbestos))); (TWA mppcf (((c) Containing less than 1% quartz; if 1% quartz or more, use quartz limit.)))
US OSHA Permissible Exposure Levels (PELs) - Table Z1	talc	Silicates (less than 1% crystalline silica): Talc (containing asbestos); use asbestos limit	Not Available	Not Available	Not Available	see 29 CFR 1910.1001; See Table Z-3
US OSHA Permissible Exposure Levels (PELs) - Table Z1	talc	Silicates (less than 1% crystalline silica): Talc (containing no asbestos), respirable dust	Not Available	Not Available	Not Available	See Table Z-3
US ACGIH Threshold Limit Values (TLV)	talc	Talc: Containing asbestos fibers	Not Available	Not Available	Not Available	Use Asbestos TLV® (K)
US ACGIH Threshold Limit Values (TLV)	talc	Talc: Containing no asbestos fibers	2 mg/m3	Not Available	Not Available	Pulm fibrosis; pulm func
US NIOSH Recommended Exposure Limits (RELs)	ferric oxide	Iron(III)oxide, Iron oxide red, Red iron oxide, Red oxide	Not Available	Not Available	Not Available	See Appendix D
US NIOSH Recommended Exposure Limits (RELs)	ferric oxide	Ferric oxide, Iron(III) oxide	5 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	ferric oxide	Iron oxide fume	10 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	ferric oxide	Rouge: Total dust	15 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	ferric oxide	Rouge: Respirable fraction	5 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	ferric oxide	Iron oxide (Fe2O3) (Inhalable fraction and vapor)	5 mg/m3	Not Available	Not Available	Pneumoconiosis
US NIOSH Recommended Exposure Limits (RELs)	carbon black	Acetylene black, Channel black, Furnace black, Lamp black, Thermal black	3.5 mg/m3	Not Available	Not Available	Ca See Appendix A See Appendix C
US OSHA Permissible Exposure Levels (PELs) - Table Z1	carbon black	Carbon black	3.5 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	carbon black	Carbon black (Inhalable particulate matter)	3 mg/m3	Not Available	Not Available	Bronchitis

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
3-iodo-2-propynyl butyl carbamate	Butyl-3-iodo-2-propynylcarbamate	3.3 mg/m3	36 mg/m3	220 mg/m3
ferric oxide	Iron oxide; (Ferric oxide)	15 mg/m3	360 mg/m3	2,200 mg/m3
carbon black	Carbon black	9 mg/m3	99 mg/m3	590 mg/m3

Ingredient	Original IDLH	Revised IDLH
talc	1,000 mg/m3	Not Available
C.I. Pigment Blue 15	Not Available	Not Available
3-iodo-2-propynyl butyl carbamate	Not Available	Not Available

Version No: 1.1 Page 5 of 10 Issue Date: 04/21/2020 Print Date: 04/21/2020

PD Stain Color Vial Brandy

ferric oxide	2,500 mg/m3	Not Available
carbon black	1.750 mg/m3	Not Available

OCCUPATIONAL EXPOSURE BANDING

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit
3-iodo-2-propynyl butyl carbamate	Е	≤ 0.01 mg/m³
Notes:	Occupational exposure banding is a process of assigning chemicals into s adverse health outcomes associated with exposure. The output of this pro range of exposure concentrations that are expected to protect worker hea	cess is an occupational exposure band (OEB), which corresponds to a

Exposure controls

Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

Personal protection







Eye and face protection

- Safety glasses with side shields.
- ► Chemical goggles.

Skin protection

- See Hand protection below
- ▶ Wear chemical protective gloves, e.g. PVC.

▶ Wear safety footwear or safety gumboots, e.g. Rubber

NOTE:

Hands/feet protection

Fig. The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

Body protection

See Other protection below

Other protection

Overalls. ▶ P.V.C.

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Light sensitive.		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Not Available	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable.
Possibility of hazardous reactions	See section 7

Version No: 1.1 Page 6 of 10 Issue Date: 04/21/2020 Print Date: 04/21/2020

PD Stain Color Vial Brandy

Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

ECTION 11 TOXICOLOGIC	AL INFORMATION				
ormation on toxicological e	ffects				
Inhaled	models). Nevertheless, good hygiene practice requires the occupational setting.	ects or irritation of the respiratory tract (as classified by EC Directives using animal at exposure be kept to a minimum and that suitable control measures be used in an fume may result in headache, cold sweat and weak pulse. Capillary, kidney, liver and poisoning.			
Ingestion	The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence. A metallic taste, nausea, vomiting and burning feeling in the upper stomach region occur after ingestion of copper and its derivatives. The vomitus is usually green/blue and discolours contaminated skin.				
Skin Contact	Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. There is some evidence to suggest that this material can cause inflammation of the skin on contact in some persons. Exposure to copper, by skin, has come from its use in pigments, ointments, ornaments, jewellery, dental amalgams and IUDs (intra-uterine devices), and in killing fungi and algae. Although copper is used in the treatment of water in swimming pools and reservoirs, there are no reports of toxicity from these applications. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.				
Еуе	characterised by tearing or conjunctival redness (as with v	Although the liquid is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised by tearing or conjunctival redness (as with windburn). Copper salts, in contact with the eye, may produce inflammation of the conjunctiva, or even ulceration and cloudiness of the cornea.			
Chronic	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result hardness of the skin, scar formation, exudation and reddischronic excessive intake of iron have been associated with over iron are at an increased risk.	ults available. Animal testing shows that skin in exposure to copper may lead to			
Chronic	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result hardness of the skin, scar formation, exudation and reddischronic excessive intake of iron have been associated with over iron are at an increased risk.	xposure through inhalation, in contact with skin and if swallowed. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control ter or mutations, but there is not enough data to make an assessment.			
Chronic PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged exit this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity resultandress of the skin, scar formation, exudation and reddisting Chronic excessive intake of iron have been associated with over iron are at an increased risk. There has been concern that this material can cause candidate.	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control per or mutations, but there is not enough data to make an assessment.			
	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result hardness of the skin, scar formation, exudation and reddist Chronic excessive intake of iron have been associated with over iron are at an increased risk. There has been concern that this material can cause cand	xposure through inhalation, in contact with skin and if swallowed. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control ter or mutations, but there is not enough data to make an assessment.			
	Toxic: danger of serious damage to health by prolonged exit this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity resultandress of the skin, scar formation, exudation and reddisting Chronic excessive intake of iron have been associated with over iron are at an increased risk. There has been concern that this material can cause candidate.	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control per or mutations, but there is not enough data to make an assessment.			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result hardness of the skin, scar formation, exudation and reddisting Chronic excessive intake of iron have been associated with over iron are at an increased risk. There has been concern that this material can cause candidate. TOXICITY Not Available	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION Not Available			
	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cance. TOXICITY Not Available TOXICITY	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION Not Available IRRITATION Eye: no adverse effect observed (not irritating)[1] Skin (human): 0.3 mg/3d-l mild			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cance to the compound of the concern that the material can cause cance to the compound of the compound o	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION Not Available IRRITATION Eye: no adverse effect observed (not irritating)[1]			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cance to the compound of the concern that the material can cause cance to the compound of the compound o	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION Not Available IRRITATION Eye: no adverse effect observed (not irritating) ^[1] Skin (human): 0.3 mg/3d-l mild			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity result hardness of the skin, scar formation, exudation and reddist Chronic excessive intake of iron have been associated with over iron are at an increased risk. There has been concern that this material can cause cance. TOXICITY Not Available TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Oral (rat) LD50: >5000 mg/kg ^[1]	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION Not Available IRRITATION Eye: no adverse effect observed (not irritating) ^[1] Skin (human): 0.3 mg/3d-I mild Skin: no adverse effect observed (not irritating) ^[1]			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cand TOXICITY Not Available TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Oral (rat) LD50: >5000 mg/kg ^[1]	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control over or mutations, but there is not enough data to make an assessment. IRRITATION IRRITATION Eye: no adverse effect observed (not irritating)[1] Skin (human): 0.3 mg/3d-I mild Skin: no adverse effect observed (not irritating)[1] IRRITATION			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddist Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cand TOXICITY Not Available TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Oral (rat) LD50: >5000 mg/kg ^[1] TOXICITY Oral (rat) LD50: >10,000 mg/kg ^[2]	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control cer or mutations, but there is not enough data to make an assessment. IRRITATION			
PD Stain Color Vial Brandy	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cand TOXICITY Not Available TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Oral (rat) LD50: >5000 mg/kg ^[1]	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control per or mutations, but there is not enough data to make an assessment. IRRITATION			
PD Stain Color Vial Brandy talc C.I. Pigment Blue 15	Toxic: danger of serious damage to health by prolonged extra this material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddist Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cance to the compound of the compound o	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control per or mutations, but there is not enough data to make an assessment. IRRITATION Not Available IRRITATION Eye: no adverse effect observed (not irritating) ^[1] Skin (human): 0.3 mg/3d-I mild Skin: no adverse effect observed (not irritating) ^[1] IRRITATION Eye (human): non-irritant Skin (human): non-irritant			
PD Stain Color Vial Brandy talc C.I. Pigment Blue 15	Toxic: danger of serious damage to health by prolonged exit This material can cause serious damage if one is exposed produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity rest hardness of the skin, scar formation, exudation and reddis Chronic excessive intake of iron have been associated wit over iron are at an increased risk. There has been concern that this material can cause cand TOXICITY Not Available TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Oral (rat) LD50: >5000 mg/kg ^[1] TOXICITY Oral (rat) LD50: >10,000 mg/kg ^[2] TOXICITY dermal (rat) LD50: >2000 mg/kg ^[2]	xposure through inhalation, in contact with skin and if swallowed. It to it for long periods. It can be assumed that it contains a substance which can be ults available. Animal testing shows that skin in exposure to copper may lead to sh changes. It damage to the liver and pancreas. People with a genetic disposition to poor control over or mutations, but there is not enough data to make an assessment. IRRITATION Rot Available IRRITATION Eye: no adverse effect observed (not irritating)[1] Skin (human): 0.3 mg/3d-I mild Skin: no adverse effect observed (not irritating)[1] IRRITATION Eye (human): non-irritant Skin (human): non-irritant IRRITATION Eye: adverse effect observed (irreversible damage)[1]			

TOXICITY

TOXICITY

Oral (rat) LD50: >10000 $mg/kg^{[2]}$

dermal (rat) LD50: >2000 $mg/kg^{[1]}$

Oral (rat) LD50: >15400 $mg/kg^{[2]}$

ferric oxide

carbon black

Legend:

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

IRRITATION

Not Available

IRRITATION

Eye: no adverse effect observed (not irritating) $^{[1]}$

Skin: no adverse effect observed (not irritating) $\ensuremath{^{[1]}}$

Version No: 1.1 Page 7 of 10 Issue Date: 04/21/2020 Print Date: 04/21/2020

PD Stain Color Vial Brandy

The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. **PD Stain Color Vial Brandy** for copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity results available. In an acute dermal toxicity study (OECD TG 402), one group of 5 male rats and 5 groups of 5 female rats received doses of 1000, 1500 and 2000 mg/kg bw via dermal application for 24 hours. The overuse of talc in nursing infants has resulted in respiratory damage causing fluid in the lungs and lung inflammation which may lead to death within hours of inhalation. Long-term exposure can also cause a variety of respiratory symptoms. TALC The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans Evidence of carcinogenicity may be inadequate or limited in animal testing. For 3-iodo-2-propynyl butyl carbamate (IPBC): 3-IODO-2-PROPYNYL BUTYL Acute toxicity studies with IPBC show low toxicity except severe eye irritation. Animal testing showed that extended exposure may cause CARBAMATE decreased weight gain and increased red cell and eosinophil counts. Inhalation (rat) TCLo: 50 mg/m3/6h/90D-I Nil reported

WARNING: This substance has been classified by the IARC as Group 2B: Possibly Carcinogenic to Humans. Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition TALC & FERRIC OXIDE known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound.

TALC & CARBON BLACK No significant acute toxicological data identified in literature search.

Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	×	Reproductivity	×
Serious Eye Damage/Irritation	×	STOT - Single Exposure	×
Respiratory or Skin sensitisation	✓	STOT - Repeated Exposure	~
Mutagenicity	×	Aspiration Hazard	×

Legend:

X − Data either not available or does not fill the criteria for classification

- Data available to make classification

SECTION 12 ECOLOGICAL INFORMATION

CARBON BLACK

Toxicity

	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURC
PD Stain Color Vial Brandy	Not Available	Not Available	Not Available	Not Available	Not Available
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURC
talc	LC50	96	Fish	89-581.016mg/L	2
	EC50	96	Algae or other aquatic plants	7-202.7mg/L	2
	NOEC	720	Crustacea 1-459.798mg/L		2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURC
	LC50	96	Fish	>3-200mg/L	2
C.I. Pigment Blue 15	EC50	48	Crustacea	>100mg/L	2
	EC50	72	Algae or other aquatic plants	>100mg/L	2
	NOEC	504	Crustacea	>1mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOUR
	LC50	96	Fish	Fish 0.067mg/L	
3-iodo-2-propynyl butyl	EC50	48	Crustacea	Crustacea 0.04mg/L	
carbamate	EC50	72	Algae or other aquatic plants	0.022mg/L	2
	EC10	72	Algae or other aquatic plants	Algae or other aquatic plants 0.0058mg/L	
	NOEC	72	Algae or other aquatic plants	Algae or other aquatic plants 0.0046mg/L	
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURC
	LC50	96	Fish	0.05mg/L	2
ferric oxide	EC50	48	Crustacea	5.11mg/L	2
	EC50	72	Algae or other aquatic plants	18mg/L	2
	NOEC	504	Fish	0.52mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURC
	LC50	96	Fish	>100mg/L	2
carbon black	EC50	48	Crustacea	>100mg/L	2
	EC50	72	Algae or other aquatic plants	>10-mg/L	2

Version No: 1.1 Issue Date: 04/21/2020 Page 8 of 10 Print Date: 04/21/2020

PD Stain Color Vial Brandy

	EC10 NOEC	72 96	Algae or other aquatic plants	>10-mg/L 2 >=1-mg/L 2
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data			

For copper:

Atmospheric Fate - Copper is unlikely to accumulate in the atmosphere due to a short residence time for airborne copper aerosols. Airborne coppers, however, may be transported over large distances

For copper: Ecotoxicity - Significant effects are expected on various species of microalgae, some species of macroalgae, and a range of invertebrates, including crustaceans, gastropods and sea urchins. Copper is moderately toxic to crab and their larvae and is highly toxic to gastropods (mollusks, including oysters, mussels and clams)

For Copper: Typical foliar levels of copper are: Uncontaminated soils (0.3-250 mg/kg); Contaminated soils (150-450 mg/kg); Mining/smelting soils (6.1-25 mg/kg80 mg/kg300 mg/kg). Terrestrial Fate: Plants - Generally, vegetation reflects soil copper levels in its foliage.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
C.I. Pigment Blue 15	HIGH	HIGH
3-iodo-2-propynyl butyl carbamate	HIGH	HIGH

Bioaccumulative potential

Ingredient	Bioaccumulation	
C.I. Pigment Blue 15	LOW (BCF = 11)	
3-iodo-2-propynyl butyl carbamate	LOW (LogKOW = 2.4542)	

Mobility in soil

Ingredient	Mobility
C.I. Pigment Blue 15	LOW (KOC = 10000000000)
3-iodo-2-propynyl butyl carbamate	LOW (KOC = 365.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

- ▶ Containers may still present a chemical hazard/ danger when empty.
- ▶ Return to supplier for reuse/ recycling if possible.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area.

- Product / Packaging disposal
- ▶ DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.

SECTION 14 TRANSPORT INFORMATION

Labels Required

Marine Pollutant NO

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

TALC IS FOUND ON THE FOLLOWING REGULATORY LISTS

Chemical Footprint Project - Chemicals of High Concern List

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Group 2B: Possibly carcinogenic to humans

US ACGIH Threshold Limit Values (Spanish)

US ACGIH Threshold Limit Values (TLV)

US AIHA Workplace Environmental Exposure Levels (WEELs)

US NIOSH Recommended Exposure Limits (RELs)

Version No: **1.1** Page **9** of **10** Issue Date: **04/21/2020**

PD Stain Color Vial Brandy

Print Date: **04/21/2020**

US NIOSH Recommended Exposure Limits (RELs) (Spanish)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US OSHA Permissible Exposure Levels (PELs) - Table Z3

US OSHA Permissible Exposure Limits - Annotated Table Z-1 (Spanish)

US OSHA Permissible Exposure Limits - Annotated Table Z-3 (Spanish)

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

C.I. PIGMENT BLUE 15 IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Clean Air Act - Hazardous Air Pollutants

US CWA (Clean Water Act) - Priority Pollutants

US CWA (Clean Water Act) - Toxic Pollutants

US EPCRA Section 313 Chemical List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

3-IODO-2-PROPYNYL BUTYL CARBAMATE IS FOUND ON THE FOLLOWING REGULATORY LISTS

US DOE Temporary Emergency Exposure Limits (TEELs)

US EPCRA Section 313 Chemical List

US List of Active Substances Exempt from the TSCA Inventory Notifications (Active-Inactive) Rule

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

FERRIC OXIDE IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US ACGIH Threshold Limit Values (Spanish)

US ACGIH Threshold Limit Values (TLV)

US AIHA Workplace Environmental Exposure Levels (WEELs)

US DOE Temporary Emergency Exposure Limits (TEELs)

US NIOSH Recommended Exposure Limits (RELs)

US NIOSH Recommended Exposure Limits (RELs) (Spanish)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US OSHA Permissible Exposure Limits - Annotated Table Z-1 (Spanish)

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

CARBON BLACK IS FOUND ON THE FOLLOWING REGULATORY LISTS

Chemical Footprint Project - Chemicals of High Concern List

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Group 2B: Possibly carcinogenic to humans

International WHO List of Proposed Occupational Exposure Limit (OEL) Values for Manufactured Nanomaterials (MNMS)

US - California Proposition 65 - Carcinogens

US - California Safe Drinking Water and Toxic Enforcement Act of 1986 - Proposition 65 List

US ACGIH Threshold Limit Values (Spanish)

US ACGIH Threshold Limit Values (TLV)

US AIHA Workplace Environmental Exposure Levels (WEELs)

US DOE Temporary Emergency Exposure Limits (TEELs)

US NIOSH Recommended Exposure Limits (RELs)

US NIOSH Recommended Exposure Limits (RELs) (Spanish)

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US OSHA Permissible Exposure Limits - Annotated Table Z-1 (Spanish)

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

SECTION 311/312 HAZARD CATEGORIES

Flammable (Gases, Aerosols, Liquids, or Solids)	
Gas under pressure	No
Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	No
Reproductive toxicity	No

Version No: 1.1 Page 10 of 10 Issue Date: 04/21/2020 Print Date: 04/21/2020

PD Stain Color Vial Brandy

No Skin Corrosion or Irritation Respiratory or Skin Sensitization Yes Serious eye damage or eye irritation No Specific target organ toxicity (single or repeated exposure) Yes Aspiration Hazard No Germ cell mutagenicity No Simple Asphyxiant No

US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)

None Reported

State Regulations

US. CALIFORNIA PROPOSITION 65

Hazards Not Otherwise Classified

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

US - CALIFORNIA PROPOSITION 65 - CARCINOGENS: LISTED SUBSTANCE

Carbon black (airborne, unbound particles of respirable size), Carbon-black extracts Listed

National Inventory Status

National Inventory	Status
Australia - AICS	Yes
Canada - DSL	Yes
Canada - NDSL	No (talc; C.I. Pigment Blue 15; 3-iodo-2-propynyl butyl carbamate; ferric oxide; carbon black)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - ARIPS	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Revision Date	04/21/2020
Initial Date	04/22/2020

CONTACT POINT

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection

OTV: Odour Threshold Value

BCF: BioConcentration Factors

BEI: Biological Exposure Index

Powered by AuthorITe, from Chemwatch.

No

^{**}PLEASE NOTE THAT TITANIUM DIOXIDE IS NOT PRESENT IN CLEAR OR NEUTRAL BASES**